

Combined Gas Law Problems

$$D \quad \frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2} \quad \frac{(12 \text{ atm})(23 \text{ L})}{200 \text{ K}} = \frac{(14 \text{ atm}) V_2}{300 \text{ K}} \quad \underline{V = 29.6 \text{ L}}$$

$$\textcircled{2} \quad \frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2} \quad \frac{(2.3 \text{ atm})(17 \text{ L})}{299 \text{ K}} = \frac{(1.5 \text{ atm}) V_2}{350 \text{ K}} \quad \underline{V = 30.5 \text{ L}}$$

$$\textcircled{3} \quad \frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2} \quad \frac{P_1 (28 \text{ L})}{318 \text{ K}} = \frac{(2.0 \text{ atm})(34 \text{ L})}{308 \text{ K}} \quad \underline{P = 2.5 \text{ atm}}$$

$$45^\circ \text{C} + 273 = 318 \text{ K}$$

$$35^\circ \text{C} + 273 = 308 \text{ K}$$

$$\textcircled{4} \quad \frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2} \quad \frac{4.5 \text{ L}}{287 \text{ K}} = \frac{V}{302 \text{ K}} \quad \underline{V = 4.7 \text{ L}}$$

$$14^\circ \text{C} + 273 = 287 \text{ K}$$

$$29^\circ \text{C} + 273 = 302 \text{ K}$$

$$\textcircled{5} \quad \frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2} \quad \frac{(88.89 \text{ atm})(17 \text{ L})}{340 \text{ K}} = \frac{P(12 \text{ L})}{367 \text{ K}} \quad \underline{P = 136 \text{ atm}}$$

$$67^\circ \text{C} + 273 = 340 \text{ K}$$

$$94^\circ \text{C} + 273 = 367 \text{ K}$$

$$\textcircled{6} \quad \frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2} \quad \frac{(0.5 \text{ atm}) V}{325 \text{ K}} = \frac{(1.2 \text{ atm})(48 \text{ L})}{320 \text{ K}} \quad \underline{V = 117 \text{ L}}$$

$$\textcircled{7} \quad \frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2} \quad \frac{(78 \text{ atm})(21 \text{ L})}{900 \text{ K}} = \frac{(45 \text{ atm}) V}{750 \text{ K}} \quad \underline{V = 30.3 \text{ L}}$$

$$\textcircled{8} \quad \frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2} \quad \frac{(5 \text{ atm})(2.9 \text{ L})}{323 \text{ K}} = \frac{(3 \text{ atm})(2.4 \text{ L})}{T} \quad \underline{T = 160 \text{ K or } -113^\circ \text{C}}$$

$$50^\circ \text{C} + 273 = 323 \text{ K}$$

$$\textcircled{9} \quad \frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2} \quad \frac{(60 \text{ atm}) V}{115 \text{ K}} = \frac{(30 \text{ atm})(29 \text{ L})}{225 \text{ K}} \quad \underline{V = 7.4 \text{ L}}$$