

p 404 9-17 odd, 18

$$\textcircled{9} \Delta Q = mc \Delta T$$

$$c = \frac{\Delta Q}{m \Delta T} = \frac{135 \times 10^3 \text{ J}}{(5.1 \text{ kg})(31.5 - 18^\circ \text{C})} = 2.0 \times 10^3 \text{ J kg}^{-1} \text{C}^{-1}$$

$$\textcircled{11} Q_{\text{loss}} = Q_{\text{gain}}$$

$$m_w c_w \Delta T_w = m_g c_g \Delta T_g$$

$$c_{\text{water}} = 4186 \text{ J kg}^{-1} \text{C}^{-1}$$

$$c_{\text{glass}} = 840 \text{ J kg}^{-1} \text{C}^{-1}$$

$$\Delta T_w = \frac{m_g c_g \Delta T_g}{m_w c_w} = \frac{(0.035 \text{ kg})(840 \text{ J kg}^{-1} \text{C}^{-1})(39.2 - 21.6^\circ \text{C})}{(0.135 \text{ kg})(4186 \text{ J kg}^{-1} \text{C}^{-1})}$$

$$\Delta T_w = 0.9156$$

$$T_{\text{initial}} = 39.2^\circ \text{C} + 0.9156 = \underline{40.1^\circ \text{C}}$$

$$\textcircled{13} Q_{\text{loss}} = Q_{\text{gain}}$$

$$m_H c_{Fe} \Delta T_H = m_w c_w \Delta T_w + m_{Fe} c_{Fe} \Delta T_{Fe}$$

$$\Delta T_H = \frac{m_w c_w \Delta T_w + m_{Fe} c_{Fe} \Delta T_{Fe}}{m_H c_{Fe}}$$

$$= \frac{(1.35 \text{ kg})(4186 \text{ J kg}^{-1} \text{C}^{-1})(25 - 20^\circ \text{C}) + (0.30 \text{ kg})(450 \text{ J kg}^{-1} \text{C}^{-1})(25 - 20^\circ \text{C})}{(0.40 \text{ kg})(450 \text{ J kg}^{-1} \text{C}^{-1})}$$

$$\Delta T_H = 160.725^\circ \text{C}$$

$$T_{\text{initial}} = 160.725 + 25^\circ \text{C} = \underline{190^\circ \text{C}}$$

$$\textcircled{15} Q = m_w c_w \Delta T_w + m_{Al} c_{Al} \Delta T_{Al}$$

$$= (0.75 \text{ kg})(4186 \text{ J kg}^{-1} \text{C}^{-1})(100 - 8^\circ \text{C}) + (0.36 \text{ kg})(900 \text{ J kg}^{-1} \text{C}^{-1})(100 - 8^\circ \text{C})$$

$$= 318642 \text{ J}$$

$$\text{Power} = \frac{\text{Energy}}{t}$$

$$t = \frac{E}{\text{Power}} = \frac{318642 \text{ J}}{750 \text{ W}} = \underline{430 \text{ s}}$$

17

$$Q_{\text{loss}} = Q_{\text{gain}}$$

$$m_{\text{Fe}} c_{\text{Fe}} \Delta T_{\text{Fe}} = m_{\text{g}} c_{\text{g}} \Delta T_{\text{g}} + m_{\text{Al}} c_{\text{Al}} \Delta T_{\text{Al}}$$

$$c_{\text{g}} = \frac{m_{\text{Fe}} c_{\text{Fe}} \Delta T_{\text{Fe}} - m_{\text{Al}} c_{\text{Al}} \Delta T_{\text{Al}}}{m_{\text{g}} \Delta T_{\text{g}}}$$

$$= \frac{(0.29 \text{ kg})(450 \text{ J kg}^{-1} \text{ C}^{-1})(180 - 38^\circ \text{C}) - (0.095 \text{ kg})(900 \text{ J kg}^{-1} \text{ C}^{-1})(38 - 10^\circ \text{C})}{(0.25 \text{ kg})(38 - 10^\circ \text{C})}$$

$$c_{\text{g}} = \underline{2300 \text{ J kg}^{-1} \text{ C}^{-1}}$$

$$c_{\text{Fe}} = 450 \text{ J kg}^{-1} \text{ C}^{-1}$$

$$c_{\text{Al}} = 900 \text{ J kg}^{-1} \text{ C}^{-1}$$

18

$$Q = mc \Delta T$$

$$E_k = \frac{1}{2} m v^2 = \frac{1}{2} (1.20 \text{ kg})(6.5 \text{ m s}^{-1})^2 = 25.35 \text{ J} = Q$$

25.35 J per hammer blow

10 hammer blows, so  $Q = 253.5 \text{ J}$

$$\Delta T = \frac{Q}{mc} = \frac{253.5 \text{ J}}{(0.014 \text{ kg})(450 \text{ J kg}^{-1} \text{ C}^{-1})} = \underline{40.2^\circ \text{C}}$$

$$c_{\text{Fe}} = 450 \text{ J kg}^{-1} \text{ C}^{-1}$$