

Write the name of the molecular compound from the given formula.

1. NO_2 nitrogen dioxide
2. CO_2 carbon dioxide
3. SO_3 sulfur trioxide
4. P_2S_5 diphosphorus pentasulfide
5. ICl_3 iodine trichloride
6. N_2O_4 dinitrogen tetroxide
7. ICl iodine monochloride
8. SiCl_4 silicon tetrachloride
9. SF_2 sulfur difluoride
10. SF_6 sulfur hexafluoride
11. NCl_3 nitrogen trichloride
12. SiO_2 silicon dioxide
13. PCl_3 phosphorus trichloride
14. XeF_4 xenon tetrafluoride
15. P_4O_{10} tetra phosphorus decoxide

Write the formula of the molecular compound from the given name.

1. Silicon Dioxide SiO₂
2. Sulfur Hexafluoride SF₆
3. Diphosphorus Pentoxide P₂O₅
4. Xenon Tetrafluoride XeF₄
5. Dinitrogen Monoxide N₂O
6. Carbon Monoxide CO
7. Dihydrogen Monoxide H₂O
8. Dinitrogen Trioxide N₂O₃
9. Dinitrogen Pentoxide N₂O₅
10. Dinitrogen Tetraoxide N₂O₄